# Study of Iron Bioavailability in a Native Nigerian Grain Amaranth Cereal for Young Children, Using a Rat Model<sup>1,2</sup>

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## ABSTRACT

Cereal Chem. 67(5):505-508

Iron bioavailability in Nigerian grain amaranth cereal fortified by two iron compounds, sodium ferric ethylenediaminetetraacetate (NaFeEDTA) and ferrous fumarate (FeC<sub>4</sub>H<sub>2</sub>O<sub>4</sub>), was compared with that in cereal fortified with ferrous sulfate (FeSO<sub>4</sub>). Grain amaranth is important because of its potential as a cereal for young children in Nigeria and other third world countries. Although hemoglobin gain in all three groups fed fortified cereal was significantly higher than that in the group fed cereal with no added iron, hemoglobin gain was highest in animals fed amaranth cereal with ferrous fumarate. Relative biological values for animals receiving unfortified amaranth cereal or cereal fortified with NaFeEDTA, ferrous fumarate, or FeSO<sub>4</sub> were 0.78, 0.93, 1.05, and 1.00,

Grain amaranth (Amaranthus caudatus), a hardy plant indigenous to the tropics, may become a future primary staple cereal crop upon which millions of people in the developing countries in Central Africa and South America will depend. It can be grown inexpensively with minimal cultivation on marginal agricultural land. As part of an effort to popularize its consumption on the West African coast, improved germ plasms were obtained from Rodale Research Inc., Kutztown, PA, and taken to Nigeria for agronomic trial plantings. Amaranth seeds have been chemically analyzed and found to contain approximately 18% protein and 8% seed oil, which indicates that amaranth is a good source of plant protein for humans (Becker et al 1981, Ologunde, unpublished data). The analysis of 35 test samples of grain amaranth from lots collected primarily from Guatemala, but also from Peru and Mexico by Bressani et al (1987a), showed an average protein content of 15% (12.8-17.4%), a net protein ratio (NPR) of 2.20, protein digestibility of approximately 80%, and a crude fiber content of 6.4%. Grain amaranth is also a rich source of minerals: 22.2 mg/100 g calcium, 47.4 mg/100 g potassium, and 249 mg/100 g phosphorus (Ologunde, unpublished data). Proteins found in most cereals, including those prepared from wheat or corn, are generally considered incomplete because they lack the essential amino acid lysine. The relatively high percentage of lysine in proteins found in grain amaranth (Marx 1977), however, makes it an effective cereal choice in developing countries where protein deficiency is a major concern.

In Peruvian maize cereal supplemented with amaranth, Morales et al (1988) found that the high protein and lipid contents of amaranth provided 9-10% of total dietary energy as fat, and 6.4-6.7% as protein, while providing only 50% of total dietary energy. In contrast, in order to provide 6.4% of protein of total dietary energy, maize had to provide 70% of total dietary energy. Particularly in the absence of dairy products in the diet, grain amaranth as a supplement or complement to common cereals

<sup>2</sup>Supported in part by AID Contract DAN-5053-G-SS-7005-00.

respectively. Body weight gain, hemoglobin gain, and concentrations of phytate and tannin as well as the protein efficiency ratio of fortified amaranth cereal were compared with the same parameters from a previous study of iron bioavailability in fortified Egyptian balady bread prepared with high-extraction wheat. Protein efficiency ratio of fortified amaranth cereal was approximately 1.6 as compared with 0.9 for the Eyptian bread. High relative biological values and expected body weight gain indicated optimum iron absorption from the amaranth cereal. This study indicates that ferrous fumarate is the iron fortifier of choice for grain amaranth cereal.

represents a uniquely beneficial protein source (Morales et al 1988).

Because of the prevalence of iron deficiency anemia in infants and young children in third world countries, the presence of bioavailable dietary iron is highly desirable. Both phytate and tannin, which interfere with iron absorption, were determined in the amaranth cereal used in this study. Dietary phytate, which has been shown to chelate metal ions (Hallberg et al 1987, Hallberg et al 1989), may also bind minerals in the gastrointestinal tract and further inhibit iron absorption (Oberleas 1983). Gillooly et al (1983) showed a significant inverse correlation between the polyphenol content of vegetables and iron absorption and observed that the presence of an individual vegetable with a strong enhancing or inhibiting effect profoundly influenced the bioavailability of nonheme iron in a mixed diet.

Although Bressani et al (1987a) investigated inhibitors and found that 33 raw Guatemalan grain amaranth test samples averaged 0.417% tannic acid and 2.05 units of trypsin inhibitor/ml of reactive mixture, the bioavailability of iron in cooked Nigerian grain amaranth has not previously been evaluated. Processing may affect the chemical form of innate and fortification iron, and the presence of ascorbic acid in the grain itself may also influence iron bioavailability. Information provided by the study is necessary to plan an intervention program among Nigerian children.

Considerable research has confirmed that iron deficiency within population groups is not only caused by inadequate iron intake but also by interactive factors that adversely affect dietary iron bioavailability (Latunde-Dada and Neale 1986). We therefore compared the phytic and tannic acid levels determined in this study with those of an earlier study conducted before the initiation of a nutritional intervention program that used fortified, highextraction (82%) wheat in Egyptian balady bread (Whittaker and Vanderveen 1990).

## **MATERIALS AND METHODS**

Grain used in the study was produced in agronomy field trials conducted on the teaching and research farm of Obafemi Awolowo University, Ile-Ife, Nigeria. Harvested grain was cleaned and cooked 7 min in four parts boiling water at atmospheric pressure with constant manual stirring in borosilicate tall-form beakers. Cooked grains were drained on soft cloth and spread thinly on absorbent paper at room temperature for about 1 hr to allow excess water to evaporate. The seeds were then transferred to a forced convection oven and maintained at 60°C for about 3 hr or until they attained a moisture content of 5%. Dried grain was ground in a motorized mortar and pestle (model MG2,

<sup>&</sup>lt;sup>1</sup>The studies reported herein were conducted according to the principles set forth in the *Guide for the Care and Use of Laboratory Animals*, Institute of Laboratory Animal Resources, National Research Council, NIH Publ. no. 85-23.

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Torsion Balance Co., Clifton, NJ) until fine enough to pass through a 60-mesh sieve. The flour was collected in polyethylene bags and stored in a freezer.

Variability in phytate and tannin content caused by processing was evaluated. The cereal was heated for 20 min at 40, 60, or 100°C, and phytate and tannin concentrations were determined and compared with those of the raw cereal. Phytate concentration was determined by a modified ion-exchange method (Harland and Oberleas 1986), and tannin content was determined by spectroscopy (Price and Butler 1977).

The relative iron bioavailability of grain amaranth cereal supplemented with one of three iron fortification compoundssodium ferric ethylenediaminetetraacetate (NaFeEDTA) (Hampene NaFe Purified Grade, W.R. Grace & Co., Nashua, NH), ferrous fumarate (FeC<sub>4</sub>H<sub>2</sub>O<sub>4</sub>) (Sigma Chemical Co., St. Louis, MO), or ferrous sulfate (FeSO<sub>4</sub>·7H<sub>2</sub>O) (Mallinckrodt Inc., Paris, KY)—was compared with that of the cereal alone in a hemoglobin repletion study with hemoglobin gain as an indicator in a rat model (Forbes et al 1989). Four groups of 10 weanling male rats of the Sprague-Dawley strain (Blue Spruce Farms, Inc., Altamont, NY) were individually housed in stainless steel cages and maintained under light- and temperature-controlled conditions. Animals received deionized, distilled water ad libitum.

Anemia was induced by phlebotomy (Tim 1979) and a lowiron diet during a seven-day depletion period. The depletion diet was the AIN 76A semipurified rat diet modified to exclude ferric citrate (American Institute of Nutrition 1980), which contained, in grams per kilogram of diet: casein 200.0, DL-methionine 3.0, cornstarch 150.0, glucose 500.0, fiber Celufil 50.0, corn oil 50.0, AIN 76 mineral mix (without ferric citrate) 35.0, AIN 76A vitamin mix 10.0, and choline bitartrate 2.0 (U.S. Biochemical Co., Cleveland, OH).

Following depletion, the animals were weighed and hematocrit and hemoglobin concentration were determined in duplicate from a specimen of fresh blood by the cyanmethemoglobin method (Crosby and Munn 1954). Animals were then randomized into four groups of approximately equal mean hemoglobin concentration and body weight.

Equal amounts of four grain amaranth-based repletion diets were prepared. Three contained approximately 35 mg Fe/kg of

diet added from one of the three experimental fortification sources; the fourth consisted of unfortified grain amaranth cereal, which contained 69 mg intrinsic Fe/kg of diet. AIN 76 mineral mix (excluding ferric citrate) and AIN 76A vitamin mix were added to all four cereal diets in proportions identical to those in the depletion diet. Iron concentrations of experimental meals were determined by atomic absorption spectroscopy (Boline and Shrenk 1977). Fresh food was weighed daily for each animal.

Body weight, hemoglobin concentration, and hematocrit percentage were determined after repletion. At the conclusion of the experiment, animals were euthanized, and the excised livers were frozen for subsequent nonheme iron determination by a slightly modified bathophenanthroline method (Torrance and Bothwell 1968, Whittaker and Vanderveen 1990). Results were expressed as both  $\mu g$  Fe/g of liver and total  $\mu g$  Fe/liver. Serum iron and total iron-binding capacity (TIBC) were measured by an electrochemical technique using a Ferrochem II serum iron/ TIBC analyzer (ESA, Inc. Bedford, MA) (Skikne 1987).

Relative biological value (RBV) was determined by comparing the hemoglobin gain in animals fed the amaranth cereal with and without fortification iron with that in animals fed cereal with added reference FeSO<sub>4</sub>.

Data were analyzed by one-way analysis of variance. Means were compared by the least significant difference (LSD) method when statistically significant (P < 0.05) (Snedecor and Cochran 1967).

#### RESULTS

Table I summarizes the comparative bioavailability of iron in fortified and unfortified grain amaranth cereals. There were no significant differences in the initial body weight, hemoglobin concentration, serum iron, TIBC, or hematocrit percentage among the four groups. Hemoglobin gain was highest in animals fed amaranth cereal with added ferrous fumarate, although hemoglobin gain in all three groups fed fortified cereal was significantly higher than in the group fed grain amaranth with no added iron. The reason that final serum iron values are low in comparison to initial values may be because all initial blood specimens were taken from unfasted animals, whereas final blood

| TABLE I<br>Comparison of Bioavailability of Iron in Fortified Grain Amaranth Cereals* |                    |                 |                 |                   |              |                  |  |
|---|--------------------|-----------------|-----------------|-------------------|--------------|------------------|--|
|   |                    |                 |                 |                   |              |                  |  |
|   |                    |                 | Ferrous         |                   | Significance |                  |  |
| Variable  | Unfortified        | NaFeEDTA        | Fumarate        | FeSO <sub>4</sub> | Р            | LSD <sup>b</sup> |  |
| Body wt, initial (g)  | $96 \pm 2^{\circ}$ | 97 ± 3          | 97 ± 3          | 96 ± 1            | 0.976        | NS               |  |
| Body wt, final (g)  | $122 \pm 4$        | $122 \pm 3$     | $124 \pm 3$     | $121 \pm 2$       | 0.889        | NS               |  |
| Body wt gain (g)  | $26 \pm 2$         | $25 \pm 1$      | $27 \pm 1$      | $25 \pm 1$        | 0.680        | NS               |  |
| Hb, <sup>d</sup> initial (g/dl)   | $5.9 \pm 0.2$      | $5.9 \pm 0.2$   | $5.9 \pm 0.2$   | $5.9 \pm 0.2$     | 0.997        | NS               |  |
| Hb, final $(g/dl)$  | $13.7 \pm 0.3$     | $15.2 \pm 0.2$  | $16.4 \pm 0.4$  | $15.9 \pm 0.3$    | <0.001       | 0.9              |  |
| Hb gain $(g/dl)$  | $7.8 \pm 0.2$      | $9.3 \pm 0.4$   | $10.5 \pm 0.5$  | $10.0 \pm 0.3$    | < 0.001      | 1.0              |  |
| Serum Fe, initial $(\mu g/dl)$  | $36 \pm 3$         | $38 \pm 3$      | $39 \pm 3$      | $44 \pm 3$        | 0.345        | NS               |  |
| Serum Fe, final $(\mu g/dl)$  | $23 \pm 5$         | $36 \pm 5$      | $84 \pm 15$     | $41 \pm 9$        | < 0.001      | 27               |  |
| TIBC, <sup>e</sup> initial (mg/dl)  | $862 \pm 26$       | $777 \pm 25$    | $782 \pm 25$    | $830 \pm 12$      | 0.067        | NS               |  |
| TIBC, final (mg/dl)   | $567 \pm 21$       | $502 \pm 12$    | $504 \pm 15$    | $535 \pm 15$      | 0.024        | 47               |  |
| Hct, <sup>f</sup> initial (%)   | $26 \pm 1$         | $27 \pm 1$      | $26 \pm 1$      | $26 \pm 1$        | 0.797        | NS               |  |
| Hct, final (%)  | $47 \pm 1$         | $53 \pm 1$      | $53 \pm 1$      | $54 \pm 1$        | < 0.001      | 2                |  |
| Hct gain (%)  | $21 \pm 1$         | $27 \pm 1$      | $27 \pm 1$      | $28 \pm 1$        | < 0.001      | 3                |  |
| Liver wt (g)  | $4.09 \pm 0.15$    | $4.00 \pm 0.09$ | $4.09 \pm 0.09$ | $3.84\pm0.07$     | 0.313        | NS               |  |
| Nonheme liver Fe $(\mu g/g)$  | $25.4 \pm 1.5$     | $43.1 \pm 6.3$  | 54.7 ± 7.8      | $31.2 \pm 3.6$    | 0.002        | 15.5             |  |
| Nonheme liver Fe ( $\mu$ g)   | $105 \pm 8$        | $172 \pm 25$    | $220\pm30$      | $121 \pm 15$      | 0.002        | 61               |  |
| RBV <sup>g</sup>  | $0.78\pm0.02$      | $0.93\pm0.04$   | $1.05\pm0.05$   | $1.00\pm0.03$     | < 0.001      | 0.10             |  |
| PER <sup>h</sup>  | 1.7 ± 0.1          | $1.6 \pm 0.1$   | $1.6 \pm 0.1$   | $1.6\pm0.1$       | 0.467        | NS               |  |

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<sup>a</sup>Ten animals per group were fed the test diets containing either 35 mg of fortification Fe/kg of diet or no fortification during the seven-day repletion period.

<sup>b</sup>Mean differences must equal or exceed the least significant difference value to be significantly different at P = 0.05. NS = not significantly different. <sup>c</sup>Mean  $\pm$  SEM (n = 10).

<sup>d</sup>Hemoglobin.

<sup>e</sup>Total iron-binding capacity.

<sup>f</sup> Hematocrit.

<sup>g</sup>Relative biological value = Hb gain of fortification compound divided by mean Hb gain of FeSO<sub>4</sub>.

<sup>h</sup>Protein efficiency ratio = Body weight gain (g)/dietary protein (g) for seven days.

|   | TABLE II                          |   |
|---|-----------------------------------|---|
| Comparison of Protein Efficiency Ratio (PER) Values i | in Egyptian Balady Bread, Grain A | maranth Cereal, and Casein-Based AIN 76A Diet |

|  |   | Egyptian Bread <sup>a</sup>   |   | Casein Diet (AIN 76A) <sup>a</sup>       |  |                                      | Grain Amaranth Cereal <sup>b</sup>    |                                      |                                       |                                       |
|--|---|---|---|--|--|--------------------------------------|---------------------------------------|--------------------------------------|---------------------------------------|---------------------------------------|
| Variable   | <b>FeSO</b> ₄   | FeSO <sub>4</sub> +<br>Na <sub>2</sub> EDTA                           | NaFeEDTA  | FeSO <sub>4</sub>                        | FeSO <sub>4</sub> +<br>Na <sub>2</sub> EDTA  | NaFeEDTA                             | Without<br>Fortification <sup>c</sup> | NaFeEDTA                             | Ferrous<br>Fumarate                   | <b>FeSO</b> ₄                         |
| Body wt gain (g)<br>Hb <sup>e</sup> gain (g/dl)<br>PER | $\begin{array}{c} 10 \pm 2^{d} \\ 5.4 \pm 0.5 \\ 0.8 \pm 0.1 \end{array}$ | $\begin{array}{c} 11 \pm 2 \\ 8.7 \pm 0.3 \\ 0.8 \pm 0.1 \end{array}$ | $   \begin{array}{r}     15 \pm 1 \\     8.7 \pm 0.3 \\     1.1 \pm 0.1   \end{array} $ | $52 \pm 2 \\ 6.0 \pm 0.2 \\ 2.2 \pm 0.1$ | $50 \pm 1$<br>$5.3 \pm 0.2$<br>$2.1 \pm 0.1$ | $53 \pm 1$<br>8.4 ± 0.4<br>2.2 ± 0.1 | $26 \pm 2$<br>7.8 ± 0.2<br>1.7 ± 0.1  | $25 \pm 1$<br>9.3 ± 0.4<br>1.6 ± 0.1 | $27 \pm 1$<br>10.5 ± 0.5<br>1.6 ± 0.1 | $25 \pm 1$<br>10.0 ± 0.3<br>1.6 ± 0.1 |

<sup>a</sup>Ten animals per group were fed test diets containing 30 mg of fortification Fe/kg during the 10-day repletion period (Whittaker and Vanderveen 1990).

<sup>b</sup>Ten animals per group were fed test diets containing either 35 mg of fortification Fe/kg of diet or no fortification during the seven-day repletion period.

<sup>c</sup> Iron level of 69 mg/kg for amaranth cereal was determined by atomic absorption spectroscopy.

<sup>d</sup>Mean  $\pm$  SEM.

<sup>e</sup>Hemoglobin.

specimens were drawn after an 18-hr fast. Nevertheless, final serum iron values were significantly higher in animals receiving ferrous fumarate than in any other group. TIBC varied inversely with iron status during the repletion period as expected. Although serum ferritin level indicates human iron status, liver nonheme iron level is the most reliable indicator of iron status in the rat. It was of special interest, therefore, that the nonheme iron level was significantly higher in animals fed cereal fortified with ferrous fumarate than in those fed grain fortified with FeSO<sub>4</sub>. The nonheme iron levels of the three groups fed cereals with fortification iron were all significantly higher (P = 0.002) than that of the group receiving unfortified grain amaranth cereal.

The RBVs were 0.78, 0.93, 1.05, and 1.00 for animals receiving unfortified amaranth cereal, and those fed cereal fortified with NaFeEDTA, ferrous fumarate, and FeSO<sub>4</sub>, respectively. The availability of intrinsic iron in the unfortified cereal was sufficient to produce an RBV of 0.78 and a weight gain approximately that of the other groups.

Table II compares the body weight gain, hemoglobin gain, and protein efficiency ratio values obtained in this study with those of a previous study of fortified Egyptian balady bread (Whittaker and Vanderveen 1990). That study tested the efficiency of NaFeEDTA and  $FeSO_4 + Na_2EDTA$  as potential fortifiers for a traditional Egyptian flat bread that has high bran and phytate contents and is typically baked at an extraordinarily high temperature (approximately 500°C), factors that may inhibit iron bioavailability (El Guindi et al 1988). The hemoglobin gain of the FeSO<sub>4</sub> reference animals is 5.4 g/dl for animals fed ground Egyptian bread and 10.0 g/dl for those receiving cereal. Fortification of the bread with NaFeEDTA produced a hemoglobin gain of 8.7 g/dl, which is 62% higher than that produced by FeSO<sub>4</sub> fortification. However, addition of NaFeEDTA to the cereal diet produced a 7% decline compared with the FeSO<sub>4</sub>-fortified cereal. El Guindi et al (1988) reported in a human study that NaFeEDTA seems to counteract the inhibitory effect of phytate in FeSO4-fortified bread. Thus, NaFeEDTA was chosen as an effective fortifier for Egyptian balady bread. The present study, however, indicated that the tannin and phytate in grain amaranth cereal did not inhibit iron absorption as profoundly as in the high-extraction (82%) wheat flour used in the Egyptian bread. Thus, EDTA fortification produced quite different results.

The phytate and tannin concentrations in ground grain amaranth processed in water at 40, 60, and 100°C for 20 min were compared with their contents in raw cereal (Table III). Processing the grain at 40°C for 20 min affected neither phytate nor tannin concentration. At 60°C, however, phytate concentration was reduced slightly, from 7.92 to 7.58 mg/g of cereal, and tannin from 0.22 to 0.15 mg catechin equivalents (CE)/100 g of cereal. Processing at 100°C reduced phytate and tannin to 3.01 mg/g and 0.06 mg CE/100 g, respectively.

Table IV represents phytate and tannin contents of amaranth cereal processed at  $100^{\circ}$  C for 7 min compared with that of dried, ground traditional Egyptian balady bread. Phytate concentrations (3.47 and 3.45 mg/g, respectively) were similar. The tannin content

TABLE III Effect of Processing on Phytate and Tannin Concentrations in Grain Amaranth Cereal

| Treatment                     | Phytate<br>(mg/g)   | Tannin<br>(mg CE/100 g)ª |
|-------------------------------|---------------------|--------------------------|
| Raw cereal                    | $7.92 \pm 0.08^{b}$ | $0.22 \pm 0.01$          |
| Processed at 40°C for 20 min  | $7.92 \pm 0.11$     | $0.22 \pm 0.01$          |
| Processed at 60°C for 20 min  | $7.58\pm0.05$       | $0.15 \pm 0.01$          |
| Processed at 100°C for 20 min | $3.01 \pm 0.02$     | $0.06 \pm 0.01$          |

 $^{a}CE = catechin equivalent.$ 

<sup>b</sup>Mean  $\pm$  SEM of triplicate determinations.

TABLE IV Comparison of Phytate and Tannin Contents in Grain Amaranth Cereal and Egyptian Balady Bread

| Component                         | Amaranth Cereal     | Egyptian Bread  |  |  |
|-----------------------------------|---------------------|-----------------|--|--|
| Phytate (mg/g)                    | $3.47 \pm 0.02^{a}$ | $3.45 \pm 0.02$ |  |  |
| Tannin (mg CE/100 g) <sup>b</sup> | $0.08\pm0.01$       | $0.17 \pm 0.01$ |  |  |
|                                   |                     |                 |  |  |

Mean  $\pm$  SEM of triplicate determinations.

 $^{b}CE = Catechin equivalent.$ 

of amaranth cereal, however, was less than half that found in Egyptian bread: 0.08 mg CE/100 g as compared with 0.17 mg CE/100 g. These results suggest that tannic acid may influence iron bioavailability differences in the two food vehicles.

## DISCUSSION

Iron deficiency is prevalent in developing countries where the diet is low in animal protein and heme iron. Dietary staples are cereal and vegetable products. Inhibitory factors commonly present in such a diet contribute significantly to a high incidence of iron deficiency anemia. Fortification of a universally available and inexpensive food staple is a practical intervention strategy for achieving long-term enhancement of iron status. For this reason, grain amaranth is a logical, nutritionally promising choice as a cereal food vehicle. It grows well under poor agricultural conditions and is inexpensive to prepare as a food. Other dietary intervention strategies have utilized local bread varieties, although the high bran and phytate contents in bread as well as high baking temperatures tend to inhibit iron absorption (El Guindi et al 1988).

In the present study, the RBV for the group fed a diet fortified with ferrous fumarate was significantly higher than that of the NaFeEDTA-fortified group. Several investigators (Layrisse and Martinez-Torres 1977, Viteri et al 1978, MacPhail et al 1981) have shown that NaFeEDTA is absorbed about twice as well as FeSO<sub>4</sub> from the same vegetable source. In a study of ethnic Indian females of South Africa, Ballot et al (1989) reported that NaFeEDTA-fortified curry powder, which was used as an intervention strategy, decreased the incidence of iron deficiency anemia from 22 to 5%. Lower tannin concentration, intrinsic ascorbic acid, and the RBV of innate iron in grain amaranth, however, seem to affect the optimum absorption of the iron fortification compounds. In a recent report, which was based on both human and animal studies, Hurrell et al (1989) proposed that ferrous fumarate or ferrous succinate be the iron fortifier in infant cereals. The same study also demonstrated the relative stability and high RBV of these compounds.

In summary, cooked amaranth cereal maintained a favorable protein efficiency ratio for a vegetable product. The relatively high protein and oil contents in grain amaranth make it a desirable weaning food (Imeri et al 1987). An additional advantage is that the leaves of grain amaranth are also nutritious (Marx 1977). Cooking the grain further enhances its nutritional qualities. In an animal study, Bressani et al (1987b) showed that although raw grain amaranth does not promote optimal growth, grain processed by wet cooking produces excellent growth. In the present study, normal body weight gain and high RBV indicated good iron absorption from the cereal alone, although animal groups receiving iron fortification showed a significant increase in final hemoglobin or hemoglobin gain in comparison with the group fed unfortified cereal. Thus, a cereal composed of ironfortified grain amaranth would be useful in an intervention strategy to combat protein-calorie malnutrition and iron deficiency anemia in infants and young children.

## LITERATURE CITED

- AMERICAN INSTITUTE OF NUTRITION. 1980. Second report of the Ad Hoc Committee on Standards for Nutritional Studies. J. Nutr. 110:1726.
- BALLOT, D. E., MACPHAIL, A. P., BOTHWELL, T. H., GILLOOLY, M., and MAYET, F. G. 1989. Fortification of curry powder with NaFe(III) EDTA in an iron-deficient population: Report of a controlled iron-fortification trial. Am. J. Clin. Nutr. 49:162-169.
- BECKER, R., WHEELER, E. L., LORENZ, K., STAFFORD, A. E., GROSJEAN, O. K., BETSCHART, A. A., and SAUNDERS, R. M. 1981. A compositional study of amaranth grain. J. Food Sci. 46:1175-1180.
- BOLINE, D. R., and SHRENK, W. G. 1977. Atomic absorption spectroscopy of copper and iron in plant materials. J. Assoc. Off. Anal. Chem. 60:1170-1174.
- BRESSANI, R., ELIAS, L. G., GONZALEZ, J. M., and GOMEZ-BRENES, R. 1987a. The chemical composition and protein quality of amaranth grain germ plasm in Guatemala. Arch. Latinoam. Nutr. 37(2):364-377.
- BRESSANI, R., KALINOWSKI, L.S., ORTIZ, M. A., and ELIAS, L. G. 1987b. Nutritional evaluation of roasted, flaked and popped A. caudatus. Arch. Latinoam. Nutr. 37(3):525-531.
- CROSBY, W. H., and MUNN, S. I. 1954. Standardized method for clinical hemoglobinometry. U.S. Armed Forces Med. J. 51:693-703.
- EL GUINDI, M., LYNCH, S. R., and COOK, J. D. 1988. Iron absorption from fortified flat breads. Br. J. Nutr. 59:205-213.
- FORBES, A. L., ADAMS, C. E., ARNAUD, M. J., CHICHESTER, C. O., COOK, J. D., HARRISON, B. H., HURRELL, R. F., KAHN, S. G., MORRIS, E. R., TANNER, J. T., and WHITTAKER, P. 1989. Comparison of in vitro, animal, and clinical determinations of iron bioavailability: International Nutritional Anemia Consultative Group

Task Force Report on Iron Bioavailability. Am. J. Clin. Nutr. 49:225-238.

- GILLOOLY, M., BOTHWELL, T. H., TORRANCE, J. D., MACPHAIL, A. P., DERMAN, D. P., BEZWODA, W. R., MILLS, W., CHARLTON, R. W., and MAYET, F. 1983. The effects of organic acids, phytates and polyphenols on the absorption of iron from vegetables. Br. J. Nutr. 49:331-342.
- HALLBERG, L., ROSSANDER, L., and SKANBERG, A. 1987. Phytates and the inhibitory effect of bran on iron absorption in man. Am. J. Clin. Nutr. 45:988-996.
- HALLBERG, L., BRUNE, M., and ROSSANDER, L. 1989. Iron absorption in man: Ascorbic acid and dose dependent inhibition by phytate. Am. J. Clin. Nutr. 49:140-144.
- HARLAND, B. F., and OBERLEAS, D. 1986. Anion-exchange method for determination of phytate in foods: Collaborative study. J. Assoc. Off. Anal. Chem. 69:667-670.
- HURRELL, R. F., FURNISS, D. E., BURRI, J., WHITTAKER, P., LYNCH, S. R., and COOK, J. D. 1989. Iron fortification of infant cereals: A proposal for the use of ferrous fumarate or ferrous succinate. Am. J. Clin. Nutr. 49:1274-1282.
- IMERI, A. G., ELIAS, L. G., and BRESSANT, R. 1987. Amaranth: A technological alternative for child feeding. Arch. Latinoam. Nutr. 37(1):147-159.
- LATUNDE-DADA, G. O., and NEALE, R. J. 1986. Review: Availability of iron from foods. J. Food Technol. 21:255-268.
- LAYRISSE, M., and MARTINEZ-TORRES, C. 1977. Fe(III)EDTA complex as iron fortification. Am. J. Clin. Nutr. 30:1166-1174.
- MACPHAIL, A. P., BOTHWELL, T. H., TORRANCE, J. D., DERMAN, D. P., BEZWODA, W. R., and CHARLTON, R. W. 1981. Factors affecting the absorption of iron from Fe(III)EDTA. Br. J. Nutr. 45:215-227.
- MARX, J. L. 1977. Amaranth: A comeback for the food of the Aztecs? Science 198:40.
- MORALES, E., LEMBCKE, J., and GRAHAM, G. G. 1988. Nutritional value for young children of grain amaranth and maize-amaranth mixtures: Effect of processing. J. Nutr. 118(1):78-85.
- OBERLEAS, D. 1983. The role of phytate in zinc bioavailability and homeostasis. Page 145 in: Nutritional Bioavailability of Zinc. G. Inglett, ed. American Chemical Society: Washington, DC.
- PRICE, M. L., and BUTLER, L. G. 1977. Rapid visual estimation and spectrophotometric determination of tannin content of sorghum grain. J. Agric. Food Chem. 25:1268-1272.
- SKIKNE, B. S. 1987. A commercial electrochemical method evaluated for measurement of iron status. Clin. Chem. 33:1645-1647.
- SNEDECOR, G. W., and COCHRAN, W. G. 1967. Pages 258-298 in: Statistical Methods, 6th ed. The Iowa State University Press: Ames, IA.
- TIM, K. I. 1979. Orbital Venous Anatomy of the Rat. Lab. Anim. Sci. 29:636-638.
- TORRANCE, J. D., and BOTHWELL, T.H. 1968. A simple technique for measuring storage iron concentrations in formalinised liver samples. S. Afr. J. Med. Sci. 33:9-11.
- VITERI, F. E., GARCIA-IBANEZ, R., and TORUN, B. 1978. Sodium iron NaFeEDTA as an iron fortification compound in Central America: Absorption studies. Am. J. Clin. Nutr. 31:961-971.
- WHITTAKER, P., and VANDERVEEN, J. E. 1990. Effect of EDTA on the bioavailability to rats of fortification iron used in Egyptian balady bread. Br. J. Nutr. 63:587-595.

[Received October 23, 1989. Revision received April 6, 1990. Accepted April 11, 1990.]