Comparison of Two Sample Preparation Procedures for Low-Temperature Scanning Electron Microscopy of Frozen Bread Dough¹

P. T. BERGLUND, 2,3 D. R. SHELTON, 2 and T. P. FREEMAN4

ABSTRACT

Cereal Chem. 67(2):139-140

New developments in low-temperature scanning electron microscopy (LT-SEM) allow samples to be examined in a frozen, fully hydrated state. We studied the effect of sample preparation procedure for LT-SEM on resulting ultrastructure of frozen bread dough. Frozen doughs that thawed

during SEM preparation exhibited a reticular pattern that was not apparent when samples remained frozen. The procedure used for preparing samples to be examined by LT-SEM is critical in determining the ultra-structural nature of frozen products.

Scanning electron microscopy (SEM) has been widely used to study the ultrastructure of food products. However, most sample preparation procedures require dehydration, freeze-drying, or exposure to fixatives or buffers that can produce artifacts, mask surface detail, or cause alteration of ultrastructure. Several researchers have examined dough ultrastructure by electron microscopy following either freeze-drying or chemical fixation (Aranyi and Hawrylewicz 1969; Moss 1972, 1974; Khoo et al 1975; Cumming and Tung 1975, 1977; Varriano-Marston 1977; Varriano-Marston et al 1980; Evans et al 1981; Junge et al 1981; Parades-Lopez and Bushuk 1982).

Water is an important structural and chemical component of frozen foods and, therefore, it is desirable to examine these products with a minimum of alteration. Cryogenic preparation of samples for low-temperature (LT)-SEM is a promising method to examine water distribution and microstructural changes within a frozen dough with SEM. By cooling the specimen in nitrogen slush prior to fracturing with a cooled knife, samples may be examined in a frozen, fully hydrated state.

Perhaps some of the differences in ultrastructure of bread doughs described in the literature have resulted from sampling technique rather than differences in formulation, mixing method, freezing parameters, and length of frozen storage. Ultrastructural characteristics may be better studied utilizing the new technique of LT-SEM if consistent procedures are rigorously maintained.

MATERIALS AND METHODS

Frozen doughs were prepared by a no-time dough formula, molded, and frozen at -23°C. Frozen doughs were sampled by the following two procedures: 1) samples were taken from the interior of frozen loaves and immediately placed on the specimen holder at room temperature (22°C), during which time there was at least a partial thawing; or 2) samples were cut from the interior of the frozen loaf with a precooled cutting tool and kept frozen with dry ice during placement on a cooled specimen holder. The dry ice was used only to keep the specimens completely frozen during the mounting procedure. Both samples were attached to the specimen holder with Tissue-Tek O.C.T. Compound (Miles Scientific, Naperville, IL) mixed with Aquadag (Tousimis Research Corp., Rockville, MD). The samples were then cryogenically prepared using the International Electron Optics EMscope SP2000A cryo-unit (I.E.O. Inc., Houston, TX). All samples were frozen by plunging the specimen holder into nitrogen

slush at -196°C in the integral freezing chamber. A shroud was closed over the frozen specimens prior to their transfer under vacuum onto the cold stage of the cryounit. A shrouded specimen holder provides further cryogenic protection against excessive warming and contamination with condensing vapors during transfer from the preparation chamber to the electron microscope, which has a cold stage maintained at about -180°C. The preparation chamber was maintained at -160°C or colder by means of vacuum and a liquid nitrogen cooled stage. The shroud on the specimen holder was opened and the sample was fractured with a cooled macroknife assembly. The fractured samples were again shrouded and transferred under vacuum onto the cryostage of the JEOL JSM 35 scanning electron microscope. Thawed samples were observed, sublimed at -65°C for 6 min on the cryostage, and photographed. Frozen samples were photographed after 2, 6, and 10 min sublimation on the cryostage. The samples were examined and photographed at 10 kilovolts without metallic coating.

RESULTS AND DISCUSSION

The two sampling techniques used to prepare doughs for LT-SEM produced frozen bread doughs with significantly different



Fig. 1. Frozen bread dough thawed during specimen mounting. The reticular pattern resulted from the redistribution of water and solutes during subsequent freezing in liquid nitrogen. Bar = $5 \mu m (2,000 \times)$.

unting procedure. Both samples were attached to colder with Tissue-Tek O.C.T. Compound (Miles perville, IL) mixed with Aquadag (Tousimis ..., Rockville, MD). The samples were then cryoared using the International Electron Optics 1000A cryo-unit (I.E.O. Inc., Houston, TX). All 100zen by plunging the specimen holder into nitrogen

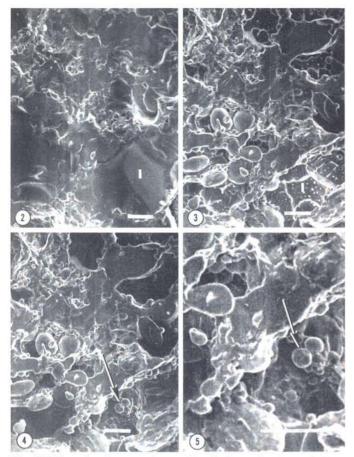
Published with the approval of the Director of the Agricultural Experiment Station, North Dakota State University, as Journal Series no. 1797.

²Graduate student and assistant professor, respectively, Department of Cereal Science and Food Technology, North Dakota State University, Fargo 58105.

³Presently assistant professor, Department of Food and Nutrition, North Dakota State University.

⁴Professor and director of Electron Microscope Laboratory, North Dakota State University.

^{@ 1990} American Association of Cereal Chemists, Inc.



Figs. 2-5. Scanning electron micrographs of a frozen dough sample kept frozen during sampling and then sublimed at -65° C. Fig. 2. Sublimed for 2 min. Ice (1) still masks much of the ultrastructure detail. Bar = $20 \ \mu m \ (600\times)$. Fig. 3. Same area of dough as in Fig. 2 but sublimed for 6 min. Only small patches of ice remain (1). Bar = $20 \ \mu m \ (600\times)$. Fig. 4. Same area as in Figs. 2 and 3 sublimed a total of 10 min. The ice has been completely removed by sublimation. Arrow indicates where ice shown in Fig. 3 has been sublimed. Bar = $20 \ \mu m \ (600\times)$. Fig. 5. Higher magnification of area shown in Fig. 4. Note the absence of the reticular pattern compared with Fig. 1. Bar = $20 \ \mu m \ (1,300\times)$.

appearances. When microscopic specimens of frozen bread doughs were prepared at room temperature, the predominant ultrastructural characteristic was the reticulate nature of the specimen (Fig. 1). This reticular pattern, best observed at magnifications of about 2,000×, is an indication of water distribution in the dough. The smooth surface of the frozen aqueous phase was replaced by a characteristic reticular pattern following etching that allowed it to be easily identified from the nonaqueous phase. According to Robards and Sleytr (1985), the reticulation is caused by sublimation of water from the aqueous phase and represents the concentration of solutes at the eutectic, the grain boundaries of ice crystals.

We found that the smaller the sample size, the greater the problem of thawing during specimen mounting at room temperature. This thawing and subsequent refreezing in nitrogen slush required for LT-SEM examination resulted in a recrystallization and a subsequent reticulation upon sublimation. Lewis and Pawley (1981) reported that ice crystal artifacts (reticula) provided a marked contrast between layers and were useful in examining water content. Crystal size was revealed by the spacing of the eutectic margins, the regions where solutes collected as they were frozen out of solution. Steere and Erbe (1977) confirmed that eutectic margins were more widely spaced in regions with the highest water content and lowest concentration of organic macromolecules.

When specimens were removed from the center of frozen loaves with a precooled cutting tool and kept frozen with dry ice for 30 sec to 1 min while being mounted onto a precooled stub, there was no evidence of a reticular network (Figs. 2–5) even when samples were sublimed at room temperature. When the frozen dough specimens were sublimed for 2 min (Fig. 2) the presence of ice still masked most ultrastructural detail. After 6 min of sublimation (Fig. 3), the starch granules and gluten matrix were more apparent. The speckled appearance on the small patches of remaining ice was probably due to solutes on the surface. An additional 4 min of sublimation (10 min total) removed the ice to a greater depth (Fig. 4). With the ice sublimed, the ultrastructure of the bread dough became apparent. No reticulation was apparent even when these specimens were examined at higher magnification (Fig. 5).

The rapid thawing that occurs when small frozen dough specimens are mounted at room temperature results in a redistribution of the water in the specimen. Subsequent freezing in liquid nitrogen causes a recrystallization of the water that forces the solutes into a reticular pattern and alters the ultrastructure of the dough. Thus, some ultrastructural differences of bread doughs described in the literature may be a result of sampling technique, rather than differences in treatments. To avoid this artifact, it is therefore essential to keep samples frozen during all stages of processing and miscroscopic examination. Studies to determine what affects the quality of frozen bread doughs are in progress.

ACKNOWLEDGMENTS

We extend a special thanks to Jay Bjerke for his assistance with microscopy and preparation of the photographs, and to Sandra Tronnes for typing this manuscript.

LITERATURE CITED

ARANYI, C., and HAWRYLEWICZ, E.J. 1969. Application of scanning electron microscopy to cereal specimens. Cereal Sci. Today 14:230. CUMMING, D. B., and TUNG, M. A. 1975. The ultrastructure of commercial wheat gluten. J. Inst. Can. Sci. Technol. Aliment. 8:67.

CUMMING, D. B., and TUNG, M. A. 1977. Modification of the ultrastructure and rheology of rehydrated commercial wheat gluten. J. Inst. Can. Sci. Technol. Aliment. 10:109.

EVANS, L. G., PEARSON, A. M., and HOOPER, G. R. 1981. Scanning electron microscopy of flour-water doughs treated with oxidizing and reducing agents. Scanning Electron Microsc. 1981(3):583.

JUNGE, R. C., HOSENEY, R. C., and VARRIANO-MARSTON, A. E. 1981. Effect of surfactants on air incorporation in dough and the crumb grain in bread. Cereal Chem. 58:338.

KHOO, U., CHRISTIANSON, D. D., and INGLETT, G. E. 1975. Scanning and transmission microscopy of dough and bread. Baker's Dig. 49(4):24.

LEWIS, E. R., and PAWLEY, J. B. 1981. Direct SEM study of frozen inner ear. Scanning 4:131.

MOSS, R. 1972. The microstructure of bread doughs. C.S.I.R.O. Food Res. Q. 32:50.

MOSS, R. 1974. Dough microstructure as affected by the addition of cysteine, potassium bromate, and ascorbic acid. Cereal Sci. Today 19:557.

PARADES-LOPEZ, D., and BUSHUK, W. 1982. Development and "underdevelopment" of wheat dough by mixing: Microscopic structure and its relations to breadmaking quality. Cereal Chem. 60:24.

ROBARDS, A. W., and SLEYTR, U. B. 1985. Pages 11-21, 147-200 in: Low Temperature Methods in Biological Electron Microscopy. vol 10. A. M. Glauert, ed. Practical Methods in Electron Microscopy. Elsevier Science Publishing Co.: New York.

STEERE, R. L., and ERBE, B. P. 1977. Comparison of acrylamide and agar gels by freeze etching. Proc. Annu. Meet. Electron Microsc. Soc. Am. 35:606.

VARRIANO-MARSTON, E. 1977. A comparison of dough preparation procedure for scanning electron microscopy. Food Technol. 31:32.

VARRIANO-MARSTON, E., HSU, K. H., and MAHDI, J. 1980. Rheological and structural changes in frozen dough. Baker's Dig. 54(1):32.